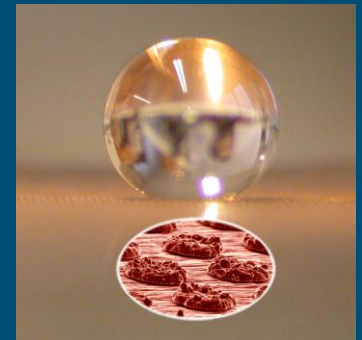


# Immersed Superhydrophobic Surfaces

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School of Science & Technology



# Overview

## 1. Functional Superhydrophobic Surfaces

- Underwater/Plastron respiration
- Superhydrophobic sol-gel mimic

## 2. Anti-fouling Microfluidic Channels

- Protein adsorption and flow enhanced detachment

## 3. Flow Enhancement in Macroscopic Pipes

- Four tube comparison of surface finish effects
- Flow enhancement visualization experiment

## 4. Drag Reduction for Settling Spheres

- Terminal velocity experiment
- Plastron drag reduction

# Functional Surfaces

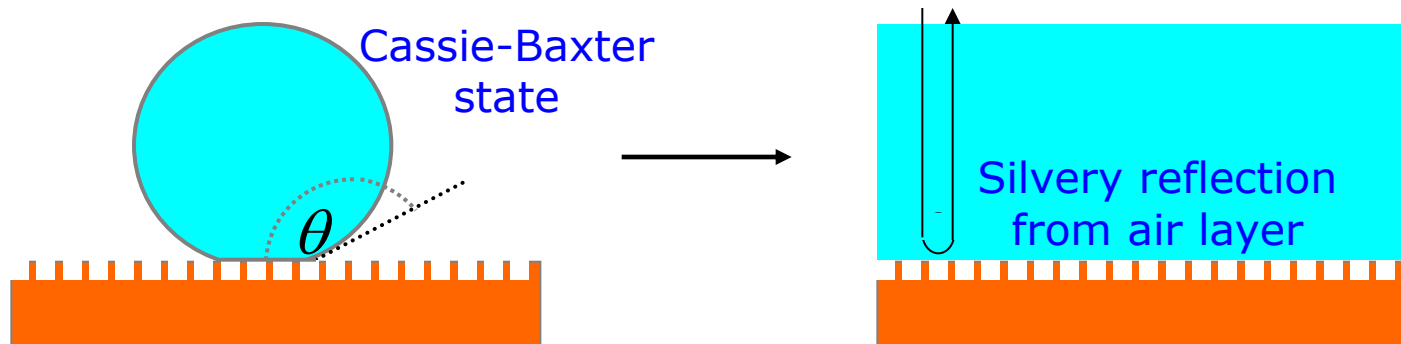
*Underwater and plastron respiration*

# Superhydrophobicity and Plastrons

## Immersed Superhydrophobic Surfaces

Provided design of features correct, penetration of water can be resisted

A silvery sheen can be seen when immersed – due to surface retained layer of air.



## Plastron Respiration

Insect physiologists have studied immersed S/H surfaces since 1940's

Silvery sheen from air layer indicates an air-water interface, e.g diving spider - acts as a gas exchange membrane to extract oxygen from water and remove carbon dioxide from insect

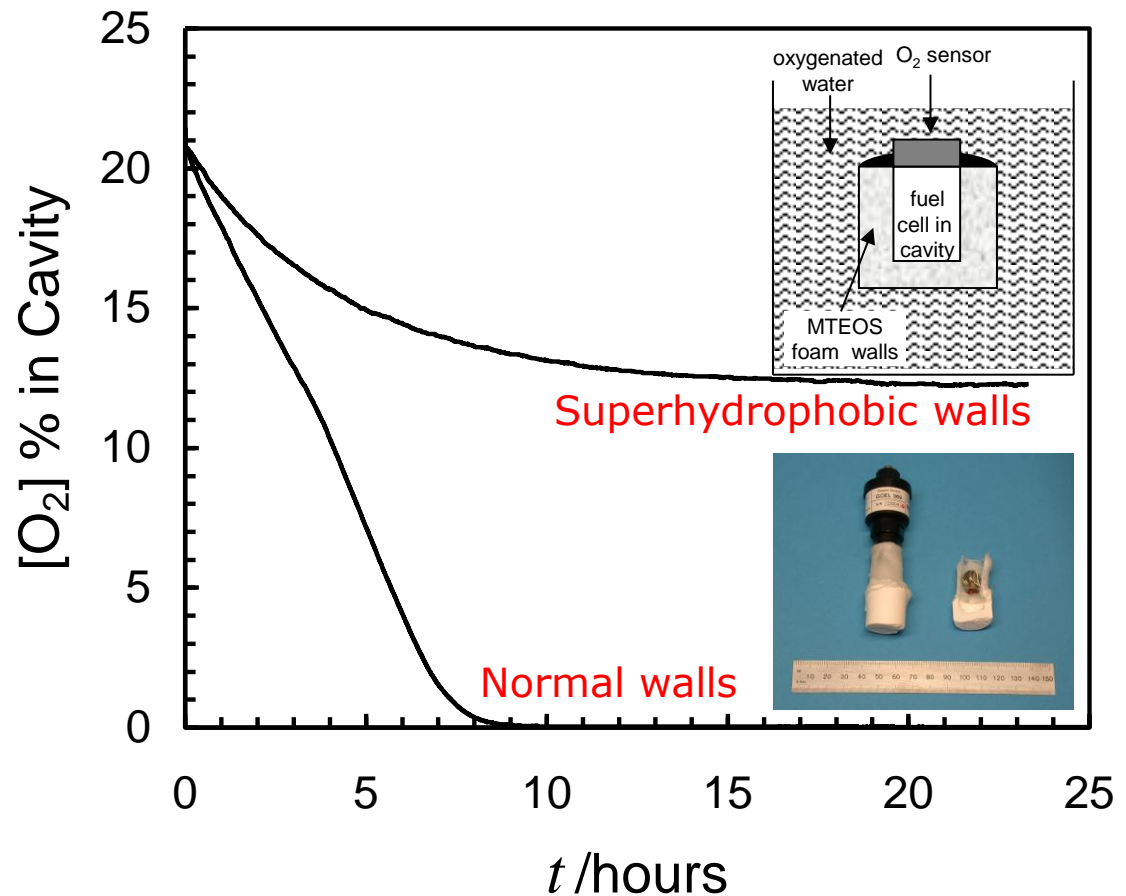
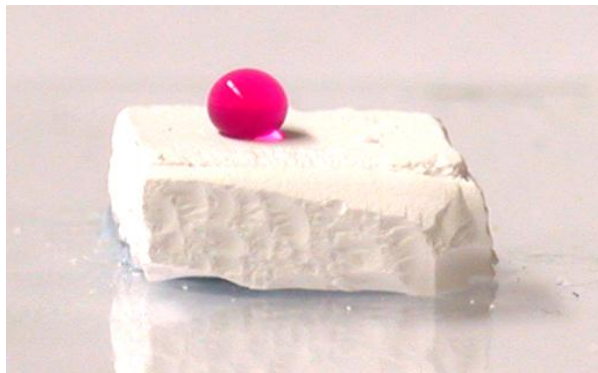
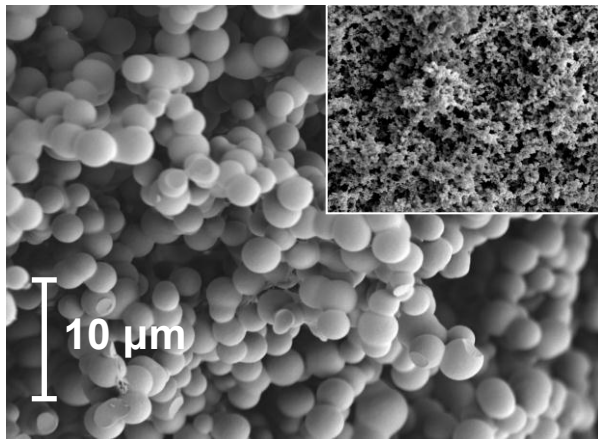


Microcosmos ©Allied Films Ltd (1996)

# Plastron/Underwater Respiration Biomimic

Intrinsically superhydrophobic MTEOS organo-silica sol-gel foam surfaces

Structure controllable from nano- to macro-porous



References Shirtcliffe *et al.*, Langmuir **19** (2003) 5626-5631; Appl. Phys. Lett. **89** (2006) art. 104106.

27 December 2013 Flynn, M.R.; Bush, J.W.M., J. Fluid Mech. **608** (2008) 275-296.

# Anti-fouling Surfaces

## *Protein Adsorption and Flow Enhanced Detachment*

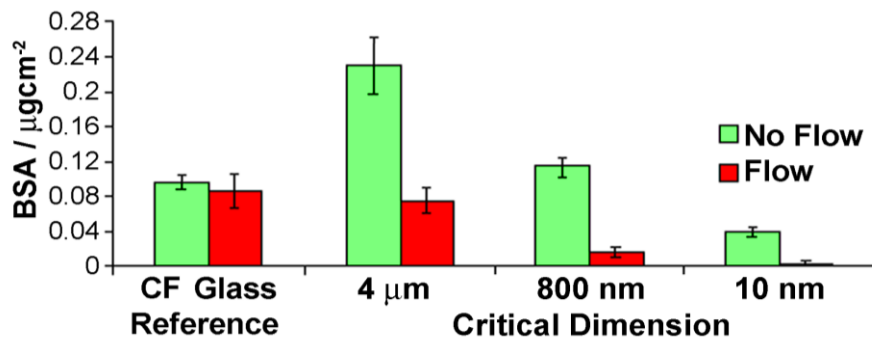
# Biofouling and Superhydrophobic Channels

## Superhydrophobic Surfaces Used

1. Glass slides
2. Sputter coated 200 nm Cu on 5 nm Ti on slides
3. Large grained (4  $\mu\text{m}$  particles, 20  $\mu\text{m}$  pores) superhydrophobic sol-gel on slides
4. Small grained (800 nm particles, 4  $\mu\text{m}$  pores) superhydrophobic sol-gel on slides
5. CuO nanoneedles (10 nm) on Cu sheet

## Proteins on Superhydrophobic Surfaces

1. Substrates incubated in BSA protein (15 nm in size) in phosphate buffer
2. Flow cell 1500 $\mu\text{m}$  x 650 $\mu\text{m}$  x 65mm using buffer solution
3. Fluorimetric assay to quantify protein removal



*Fluorinated nanoscale superhydrophobic surfaces showed almost complete removal of protein under shear flow*

Reference Koc, Y. *et al.*, Lab on a Chip 81 (2008) 582-586.

# Flow Enhancing Surfaces

## *Superhydrophobic Tubes*

Acknowledgement

Dr Yong Zhang and Dr Neil Shirtcliffe

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8





# Flow in Pipes with Superhydrophobic Walls

## Concept



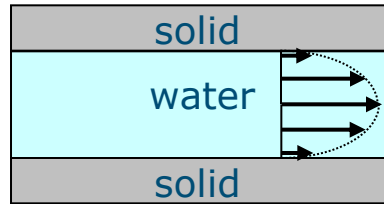
## Experiment

Forced flow through small-bore Cu tubes

Electron microscope images of hydrophobic nano-ribbon ( $1\mu\text{m} \times 100\text{nm} \times 6\text{nm}$ ) decorated internal copper surfaces of tubes (0.876 mm radii).

Side-profile optical images of droplets of b) water, and c) glycerol on surface shown in a) the original surface is shown in d)

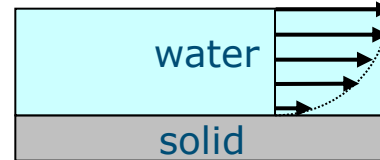
## Closed-channel



Two walls cause frictional drag

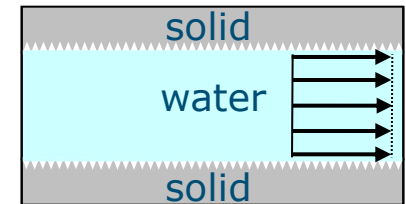
## Open-channel

Low frictional drag to air

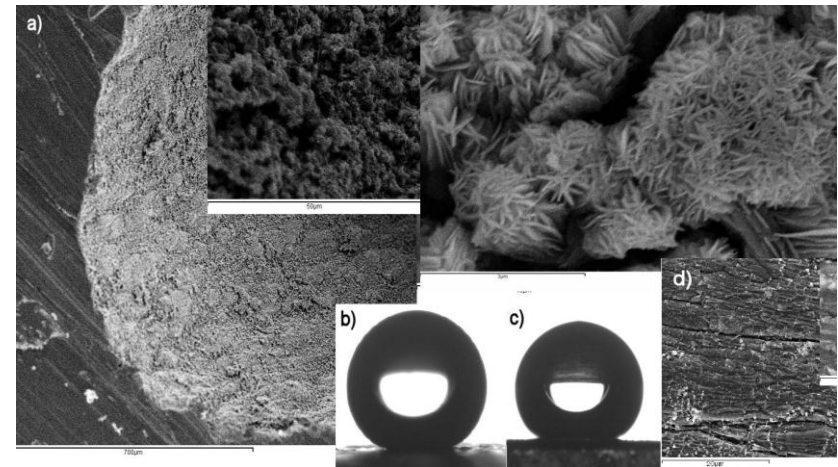


High frictional drag to solid

## Super-channel



Walls appear as cushions of air



Reference

Shirtcliffe, N.J., et al.. ACS Appl. Mater. Interf., 1 (2009) 1316-1323.

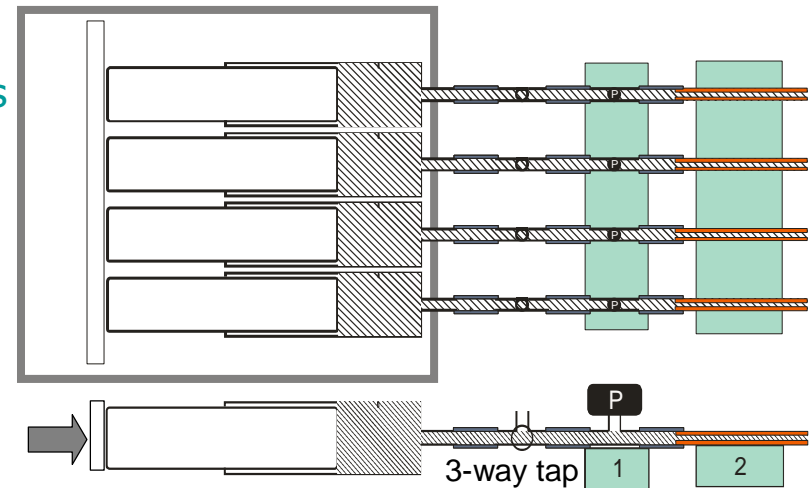
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9

# Flow Experiments

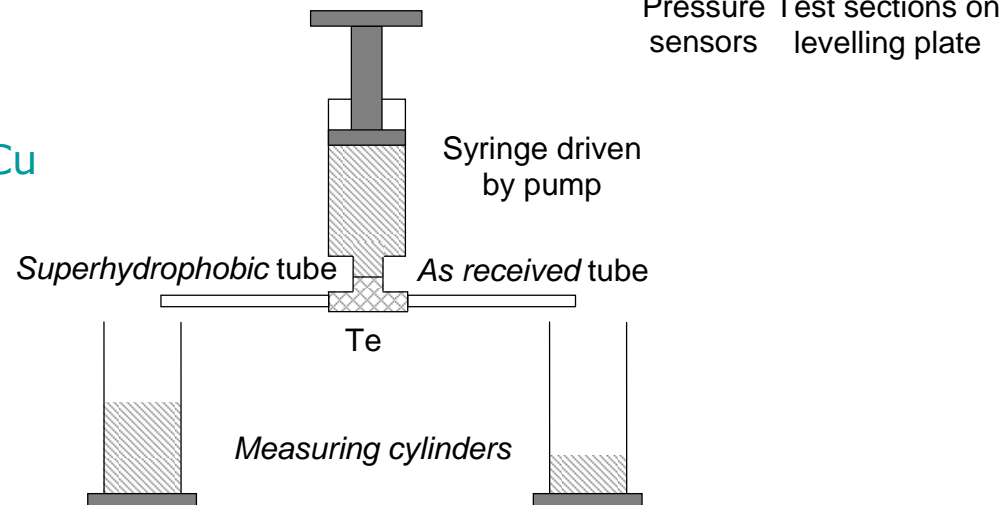
## Quantitative Experiment

1. 4 parallel tubes with 4 surface finishes
2. Cu, hydrophobic Cu, nanoribbon Cu and hydrophobic nanoribbon Cu
3. Syringe pump to force flow in all 4
4. Measure pressure drop across each



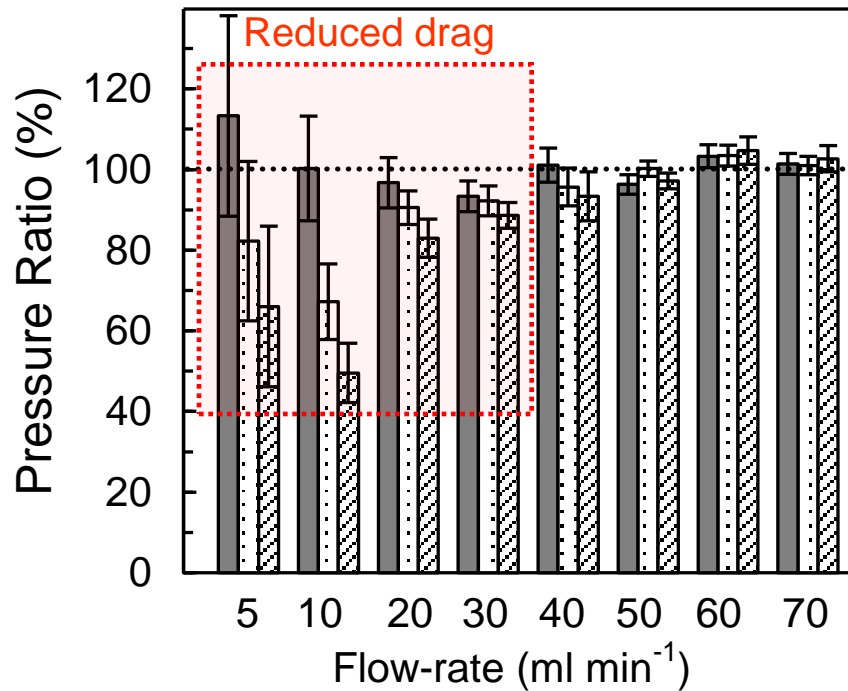
## Visualization Experiment

1. 2 tubes in T-arrangement
2. Cu and hydrophobic nanoribbon Cu
3. Syringe pump to force flow
4. Outlet volumes collected

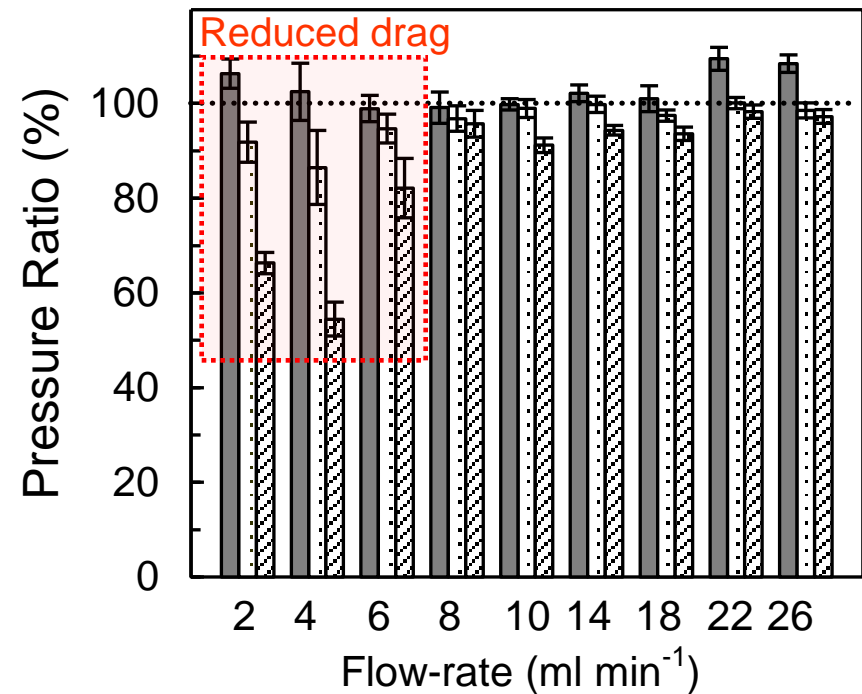


# Quantitative Results

Water



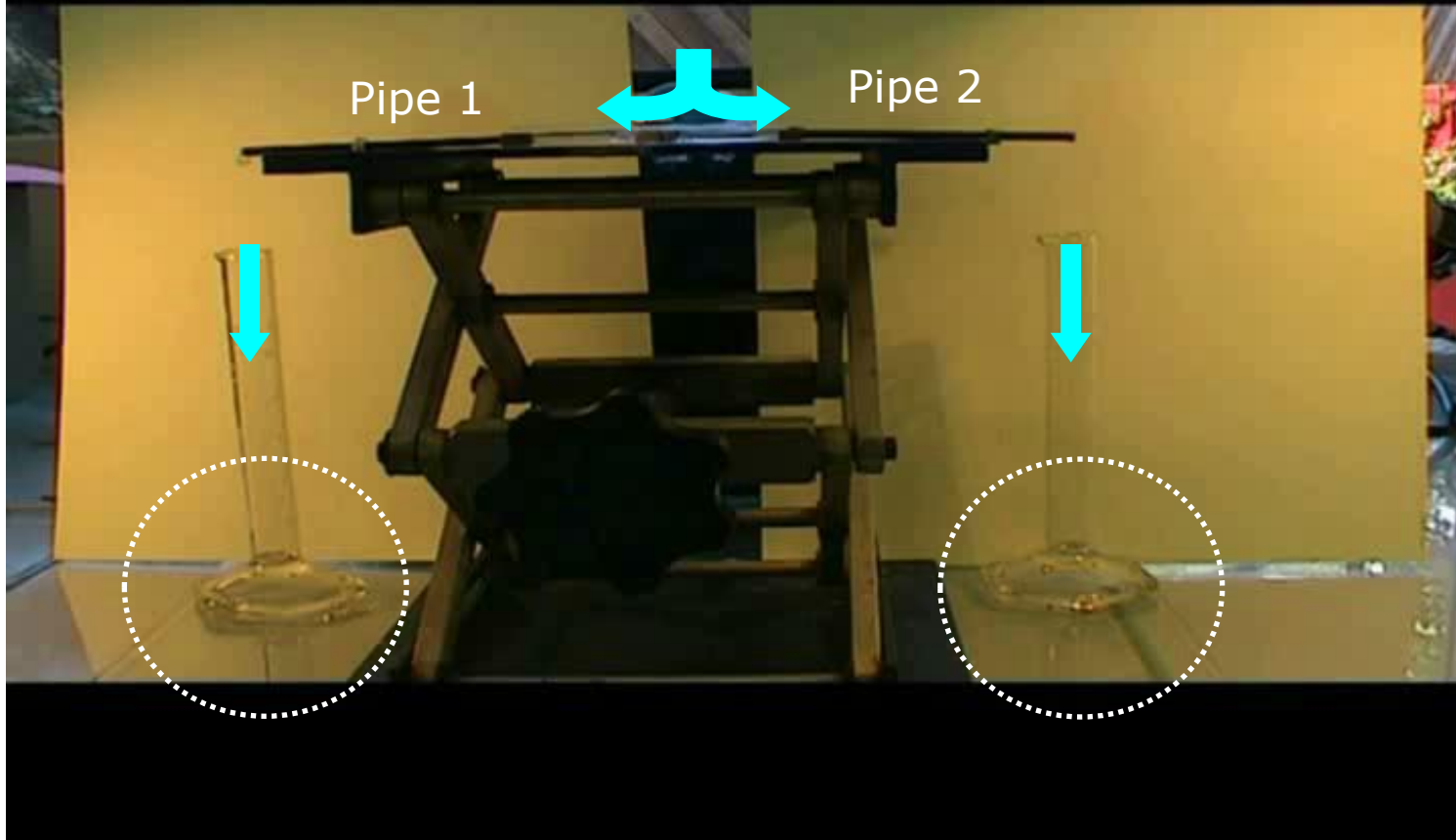
Water-Glycerol (50%)



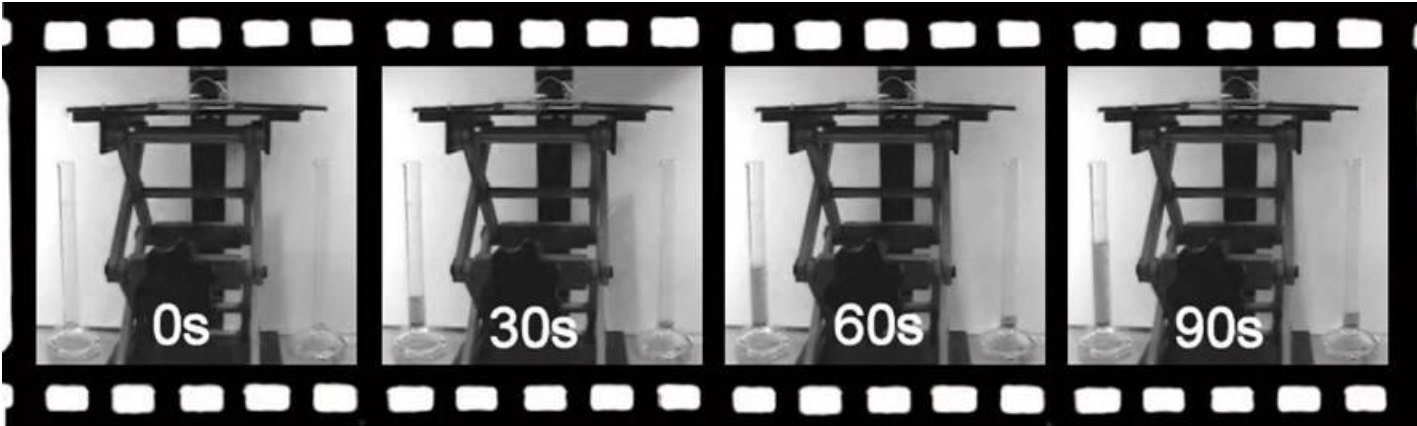
*Copper tubes with superhydrophobic inner surfaces show significantly increased flow-rates*

# Visualization Results – Set-up and Video

Two horizontal pipes –inside walls of one are coated with superhydrophobic nano-ribbons



# Visualization Results – Extracted Frames

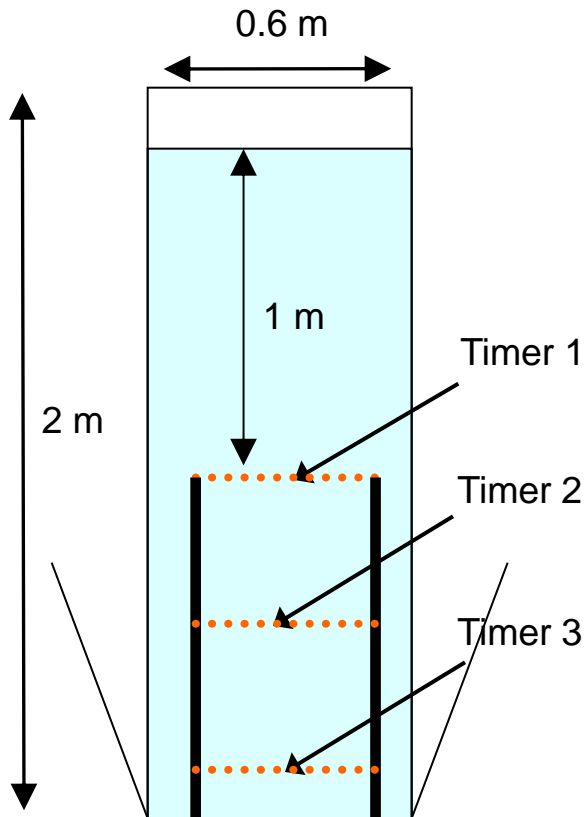


# Drag Reduction

## *Terminal velocities of settling spheres*

# Terminal Velocity Experiments

*In the presence of a fluid, a falling object eventually reaches a terminal velocity. Textbooks tell us that in water the terminal velocity does not depend on the surface chemistry .... But is that true?*

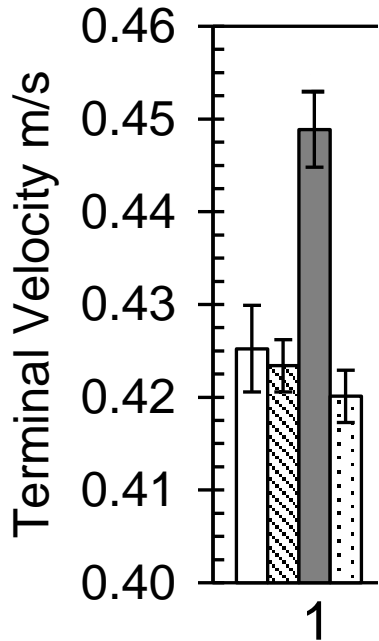


Solid sphere  
Plastron bearing sphere  
Same sphere



# Terminal Velocity Results

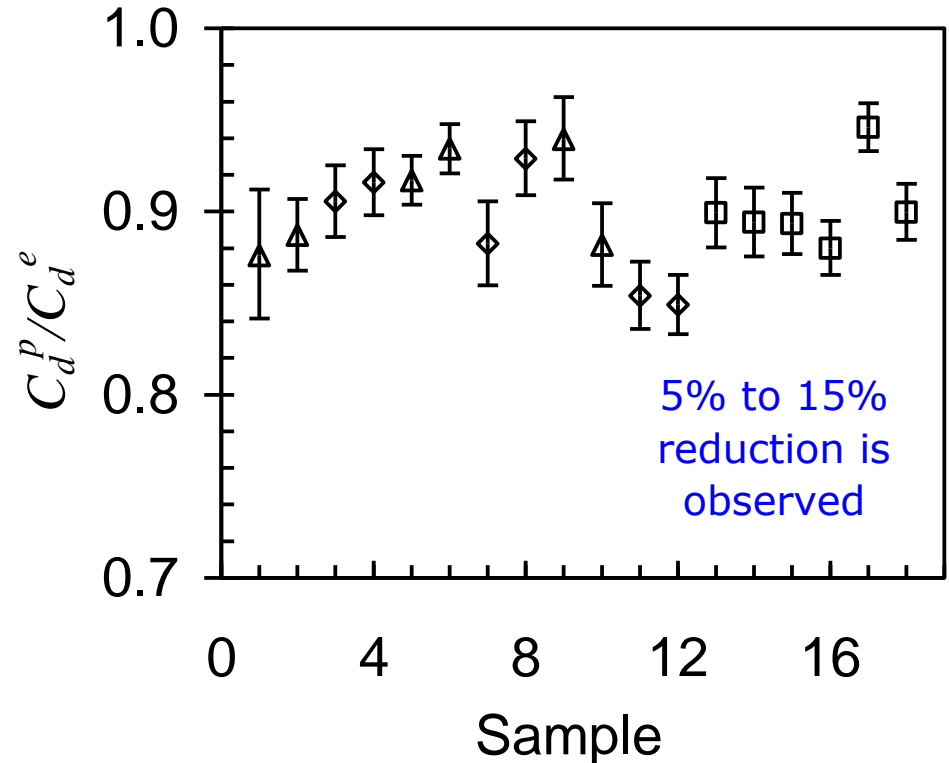
## Results for 1-inch Diameter Sphere



### Sequence of Four Bars

1. Blank surface
2. Sieved sand surface
3. (Super) Hydrophobic sand
4. Hydrophobic sand with ethanol pre-treatment to prevent plastron

## Reduction in Drag Coefficient



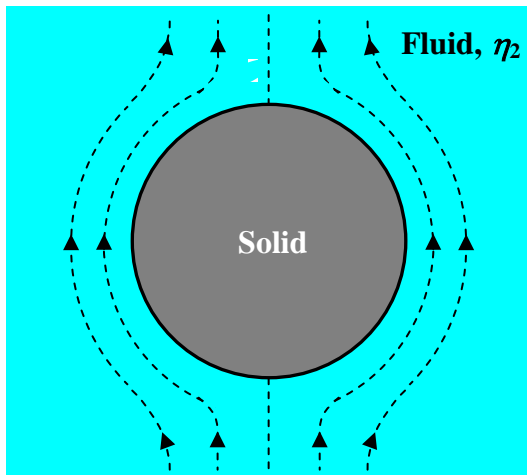
*Superhydrophobicity alone is not enough. Also need a plastron to persist to achieve drag reduction*



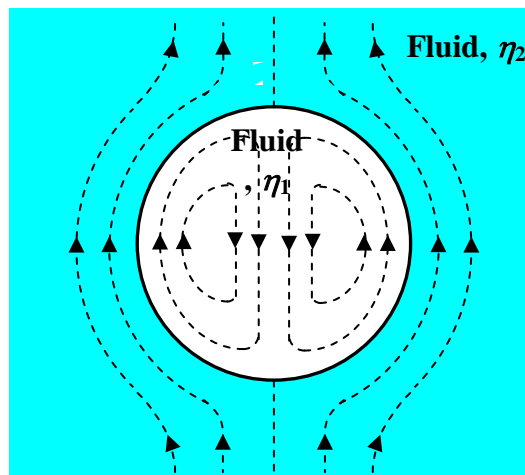
# Drag Reduction – Boundary Conditions

*Fundamental boundary condition is not "no-slip", but is continuity of shear stress*  
*Well-known drag reduction effects for gas bubbles with non-rigid interfaces in water*

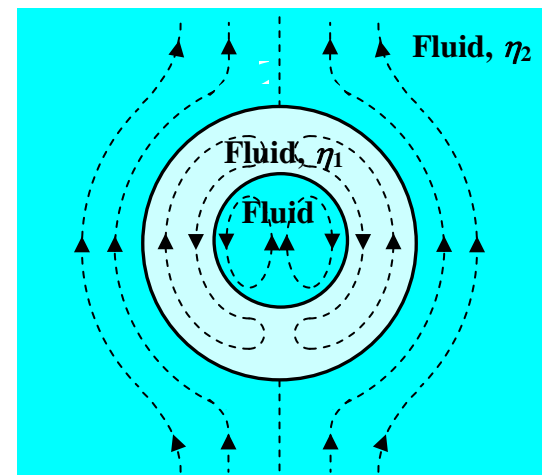
Stokes Drag (Low Re)



Hadamard-Rybczynski



Encapsulated Droplet



*Hadamard-Rybczynski drag is 25% less than Stokes drag*

# Exact Solution – Plastron Bearing Sphere

- Exact analytical solution for low Reynolds number ( $Re < 1$ ) for an encapsulated droplet has been derived by Rushton and Davies
- Apply to a solid encapsulated in air and placed in water
- Relate frictional drag of plastron bearing sphere of radius  $b$ , to Stokes drag of the sphere without a plastron

$$F_d^{SH} = \frac{2}{3} \left[ \frac{1 + 3\eta_{12}F(\varepsilon)}{1 + 2\eta_{12}F(\varepsilon)} \right] F_d^{St}(a) \quad \text{with} \quad F(\varepsilon) = \frac{(1 + \varepsilon)(2\varepsilon^2 + \varepsilon + 2)}{(1 - \varepsilon)(4\varepsilon^2 + 7\varepsilon + 4)}$$

where  $\eta_{12} = \eta_{\text{air}}/\eta_{\text{water}}$  and  $a = b/\varepsilon$  is the radius of the solid together with the sheathing film of air (i.e.  $\varepsilon < 1$  for plastron bearing sphere) and the plastron thickness is  $h(\varepsilon) = (1 - \varepsilon)b/\varepsilon$ .

- Competing effects: Air lubricates motion, but also increases cross-section of the compound object  $\Rightarrow$

*There is an optimum plastron thickness that minimizes drag*

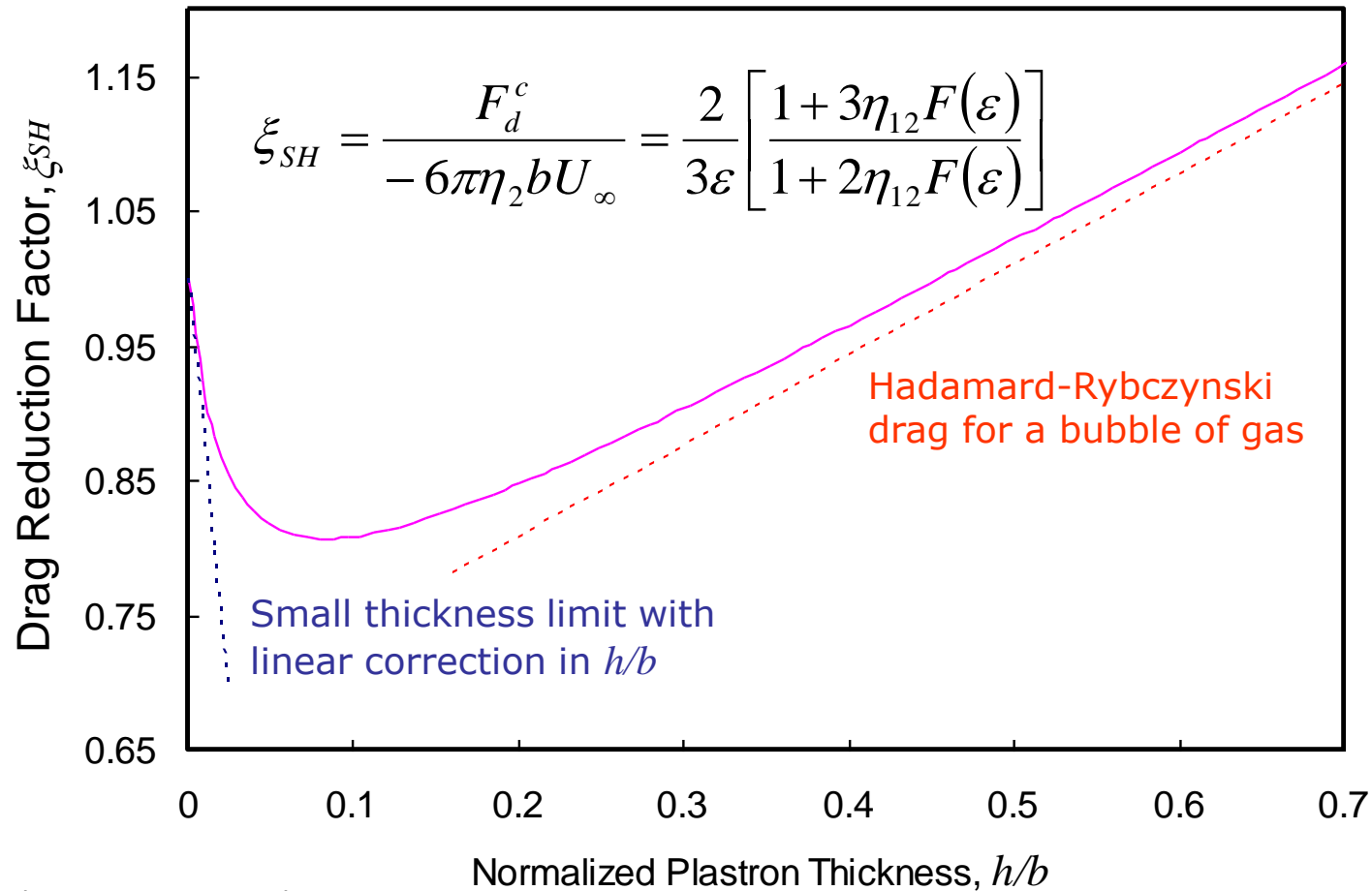
Acknowledgement Dr Morris Flynn

Reference  
27 December 2013

Rushton, E.; Davies, G. A. Int. J. Multiph. Flow 9 (1983) 337-342.  
McHale *et al.*, to be submitted.

# Drag Reduction

- Analytical solution for drag correction factor,  $\xi_{SH}$ , on the sphere is,

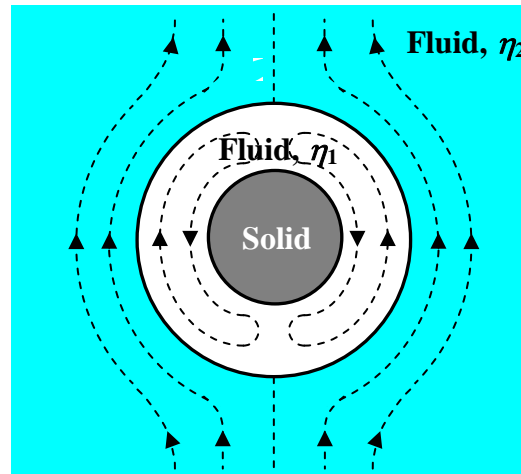


Acknowledgement Dr Morris Flynn

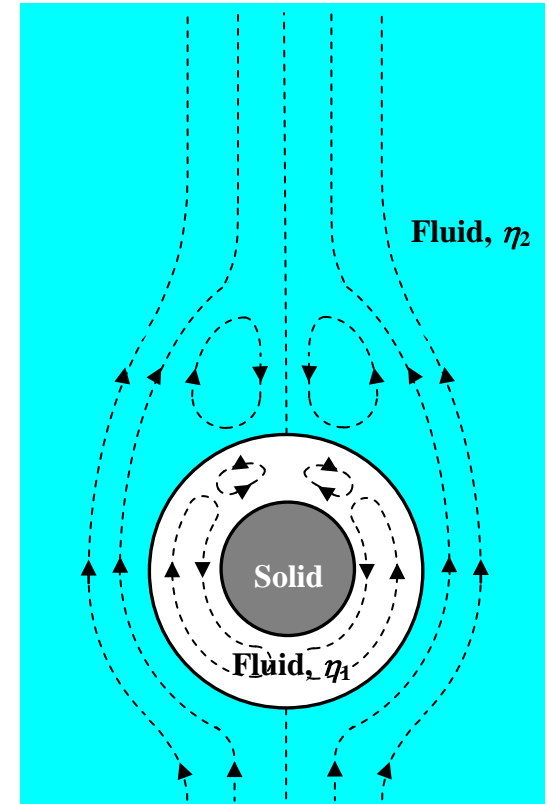
Reference McHale *et al.*, to be submitted.  
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# Low and Intermediate $Re$ numbers?

At low  $Re$ , a plastron/air layer may cause a Hadamard-Rybczynski effect and reduce drag



At intermediate  $Re$ , an air layer is likely to alter flow patterns and modify wake separation – possibly reducing drag



*A persistent plastron/air layer can give drag reduction*

# Conclusions

## 1. Immersed Superhydrophobic Surfaces can be Functional

- Oxygen can be extracted from water
- Underwater/plastron respiration is possible

## 2. Biofouling can be Reduced

- Protein adsorption can be reduced
- Flow induced detachment can be enhanced

## 3. Flow through Macroscopic Tubes can be Enhanced

- Visible enhancement is possible

## 4. Drag can be Reduced

- Plastron like boundary layers of air can lubricate flow and reduce drag
- There is an optimum length scale that is needed

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The End

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# Acknowledgements

## Collaborators

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PDRA's

Dr Carl Evans, Dr Paul Roach, Dr Yong Zhang,

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GR/S34168/01 – Electrowetting on superhydrophobic surfaces

EP/D500826/1 & EP/E043097/1 – Slip & drag reduction

EP/E063489/1 – Exploiting the solid-liquid interface

EU COST Action P21 - Physics of droplets



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